

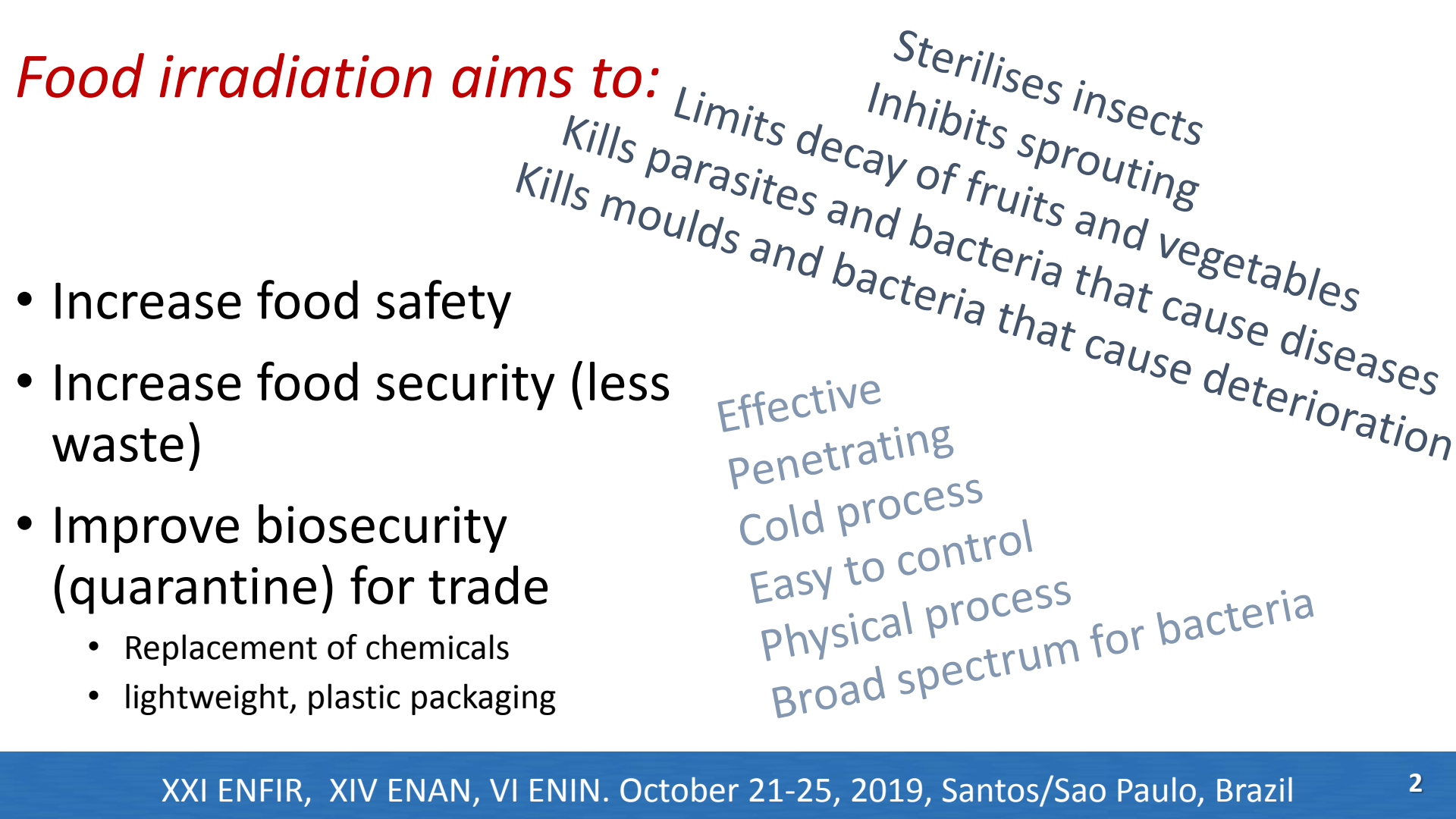


	Université		
		de Strasbourg	

Safety and Quality of Irradiated Foods

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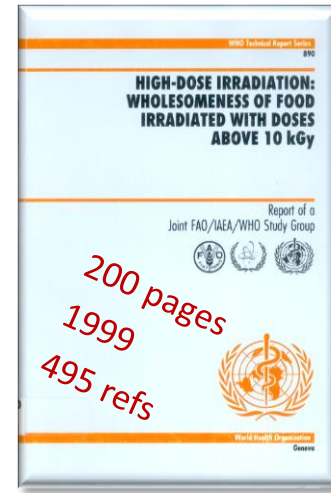
Food irradiation aims to:

- Increase food safety
- Increase food security (less waste)
- Improve biosecurity (quarantine) for trade
 - Replacement of chemicals
 - lightweight, plastic packaging

Limits decay of fruits and vegetables
Sterilises insects
Inhibits sprouting
Kills parasites and bacteria that cause diseases
Kills moulds and bacteria that cause deterioration

Effective
Penetrating
Cold process
Easy to control
Physical process
Broad spectrum for bacteria

Food irradiation should be safe if we believe:



International organizations

- FAO/WHO Joint Expert Committee (1980)
- *Codex Alimentarius (1983)*
- WHO (1992)
- WHO/FAO (1999)
- EFSA (2001)
- *Codex Alimentarius (2003)*
- EFSA (2011)

National organizations

- Food and Drug Administration (USA)
- Anses (France)
- ...

Basis of radiation chemistry

Reaction	(Hart, 1972)	k (M ⁻¹ x s ⁻¹)	Reaction	(Hart, 1972)	k (M ⁻¹ x s ⁻¹)
e ⁻ _{aq} + e ⁻ _{aq} → H ₂ + 2 OH ⁻		0.6 x 10 ¹⁰	e ⁻ _{aq} + O ₂ → •O ₂ ⁻		2.0 x 10 ¹⁰
e ⁻ _{aq} + H [•] → H ₂ + OH ⁻		2.5 x 10 ¹⁰	H [•] + H [•] → H ₂		2.0 x 10 ¹⁰
e ⁻ _{aq} + OH [•] → OH ⁻		3.0 x 10 ¹⁰	OH [•] + OH [•] → H ₂ O ₂		0.6 x 10 ¹⁰
e ⁻ _{aq} + H ₂ O → H [•] + OH ⁻		16.0	H [•] + OH [•] → H ₂ O		~ 2 x 10 ¹⁰
H [•] + OH ⁻ → e ⁻ _{aq}		1.8 x 10 ⁷	OH [•] + H ₂ → H ₂ O + H [•]		4.5 x 10 ⁷
e ⁻ _{aq} + H ₃ O ⁺ → H [•] + H ₂ O		2.06 x 10 ¹⁰	H [•] + H ₂ O ₂ → OH [•] + H ₂ O		9.0 x 10 ⁷
e ⁻ _{aq} + H ₂ O ₂ → OH [•] + OH ⁻		1.2 x 10 ¹⁰	OH [•] + H ₂ O ₂ → H ₂ O + HO [•] ₂		4.5 x 10 ⁷
			H ₃ O ⁺ + OH ⁻ → 2 H ₂ O		1.43 x 10 ¹¹

H₂O₂ and H₂ are largely consumed
 Low yields of production even if high dose is employed

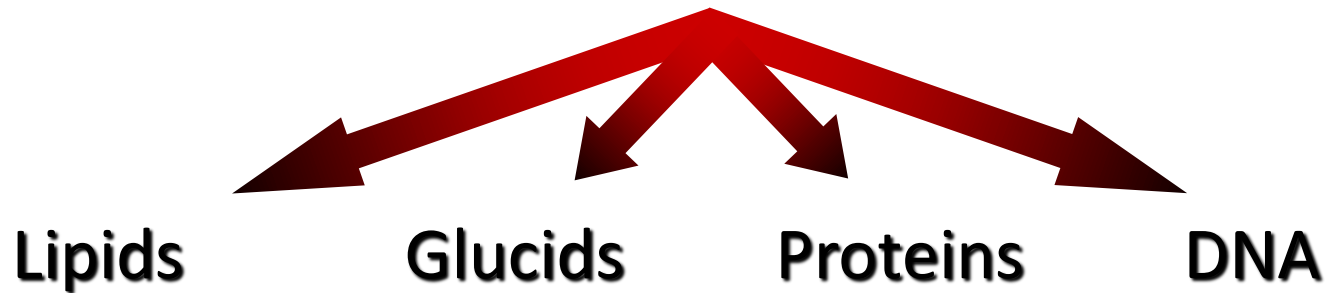
Radicals are the very first radiolysis species formed

To resume, we got : H[•] ; OH[•]

Basis of radiation chemistry

Radicals and excited species are the primary radiolytic products

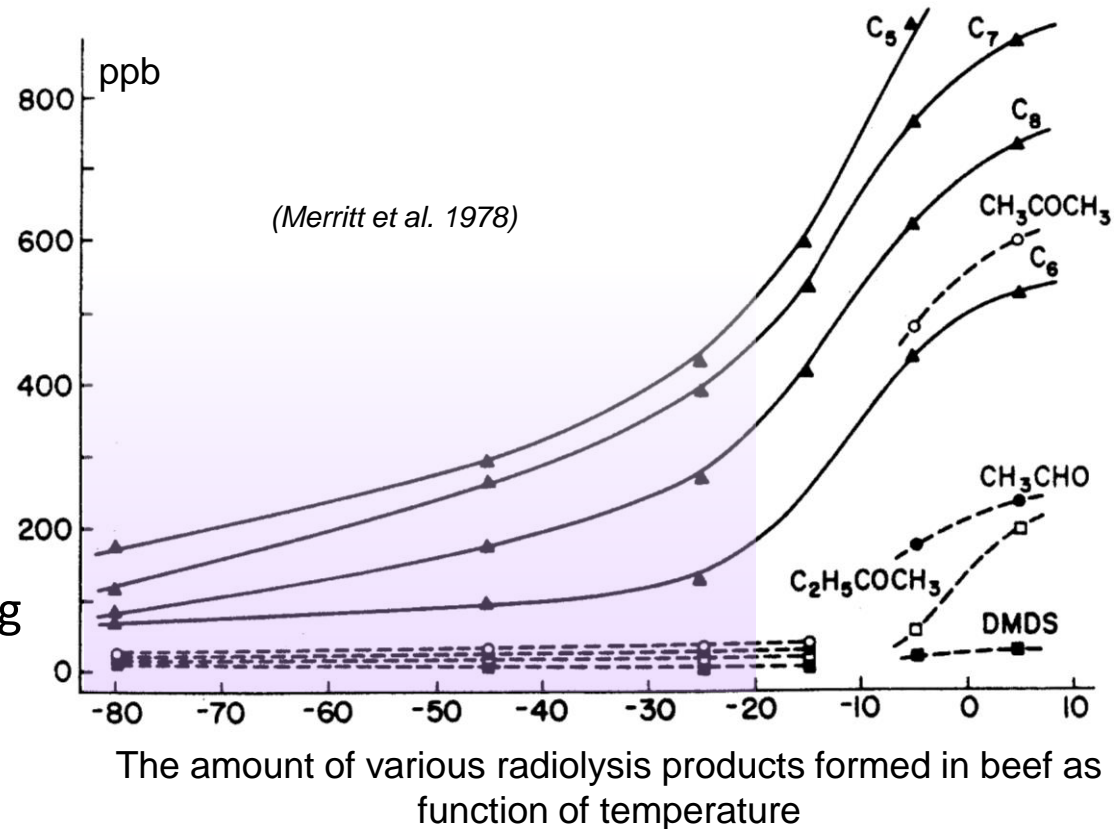
They will react with the food matrix :



Radiolytic products were created!

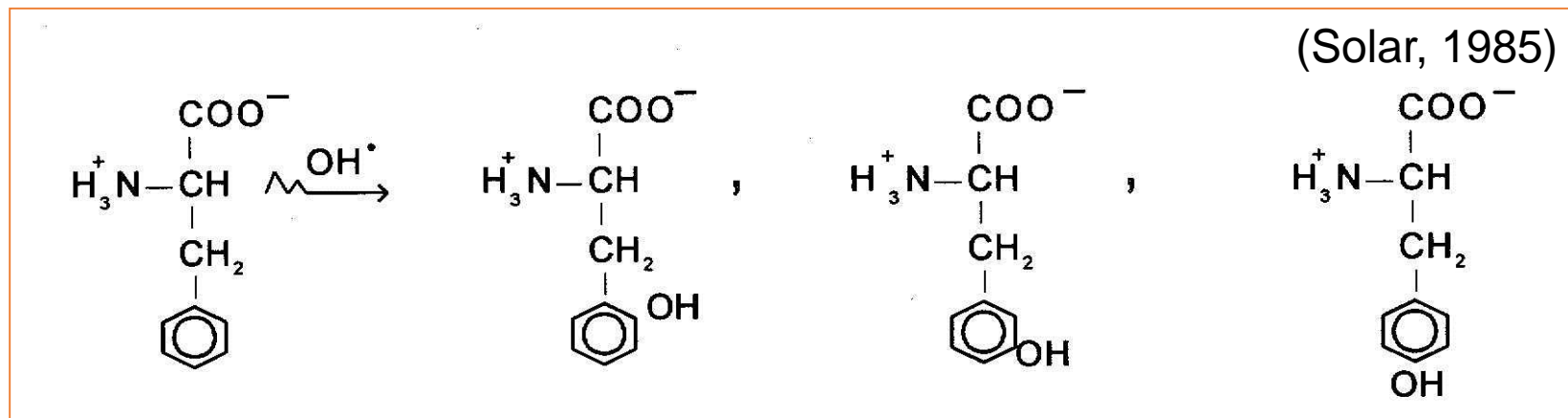
Chemical changes following irradiation at low temperature are much less than at ambient temperature or chilled.

High dose irradiation treatments should be performed at subfreezing temperatures (and in anoxic conditions)



Radiolytic products were created!

In 1983 Simic et al claimed that O-tyrosine was an URP..... But later (1990) he recovered this compound in non irradiated foods.



phenylalanine

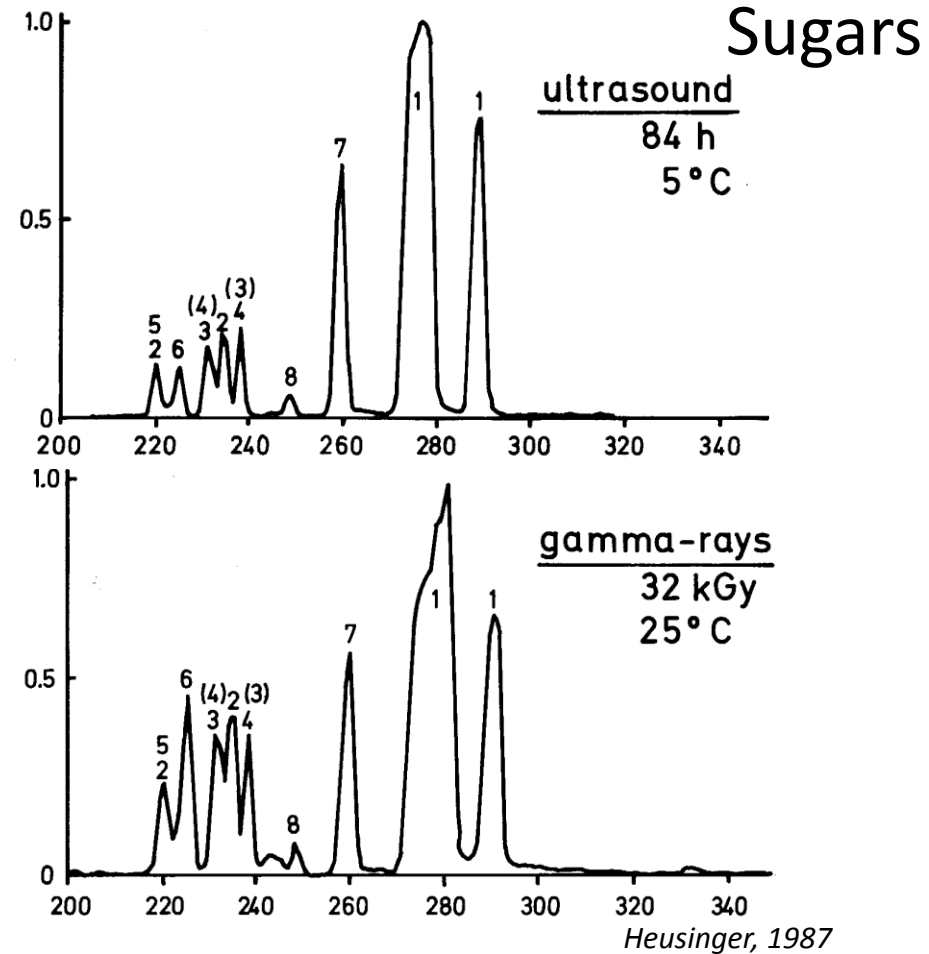
o-tyrosine
(50 %)

m-tyrosine
(14%)

p-tyrosine
(30 %)

Radiolytic products were created!

Similar reactions products are formed after radiolysis and ultrasound treatment on carbohydrates but also on proteins and DNA



Radiolytic products were created!

Radiation dose to inactivate 90% of some pathogenic micro-organism

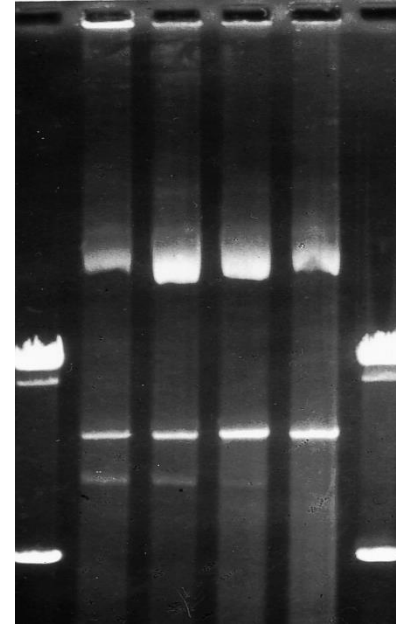
Organism	Substrate	Temp. °C	D ₁₀ (kGy)
<i>Aeromonas hydrophila</i>	in beef	2	0.14 - 0.19
<i>Campylobacter jejuni</i>	in turkey	0 - 5	0.19
<i>E.Coli 0157 : H 7</i>	in beef	5	0.27
<i>Listeria monocytogenes</i>	in chicken	2 - 4	0.77
<i>Salmonella spp.</i>	in chicken	2	0.38 - 0.77
<i>Staphylococcus aureus</i>	in chicken	0	0.36

*Radiolytic products
were created!*

Beef liver DNA extract
16 000 bp

Open circular shape →
Linear shape →
Supercoiled shape →

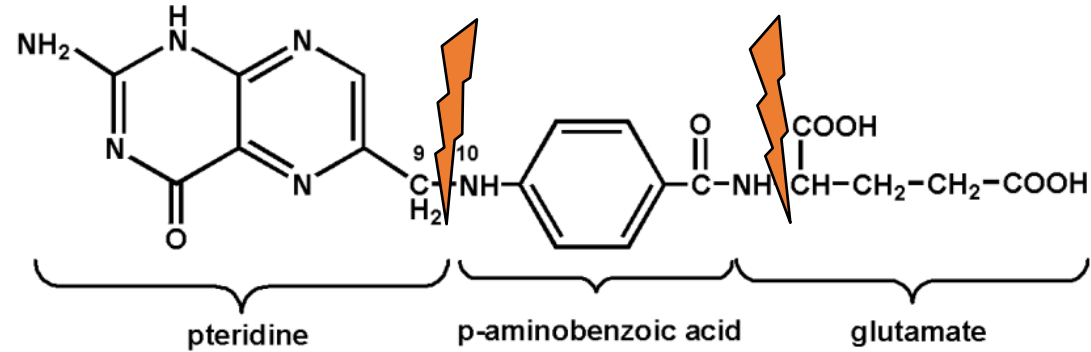
0 1 3 5 kGy



M. Bergaentzlé et al., 1994, Food Sci. Technol. Today, 8, 111-113

E. Marchioni et al., 1996, Detection Methods for Irradiated Foods - Current Status, 355-366

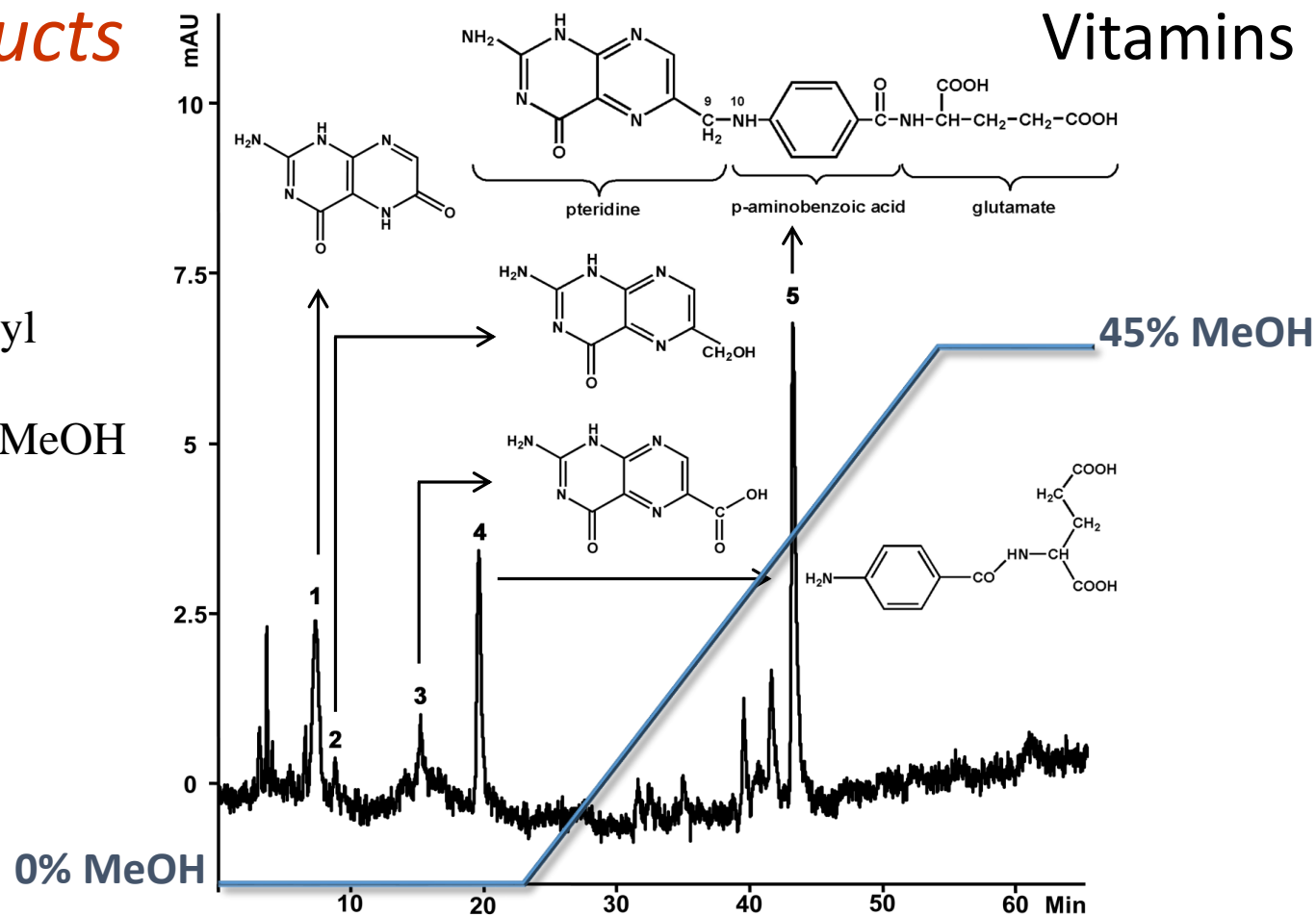
*Radiolytic products
were created!*



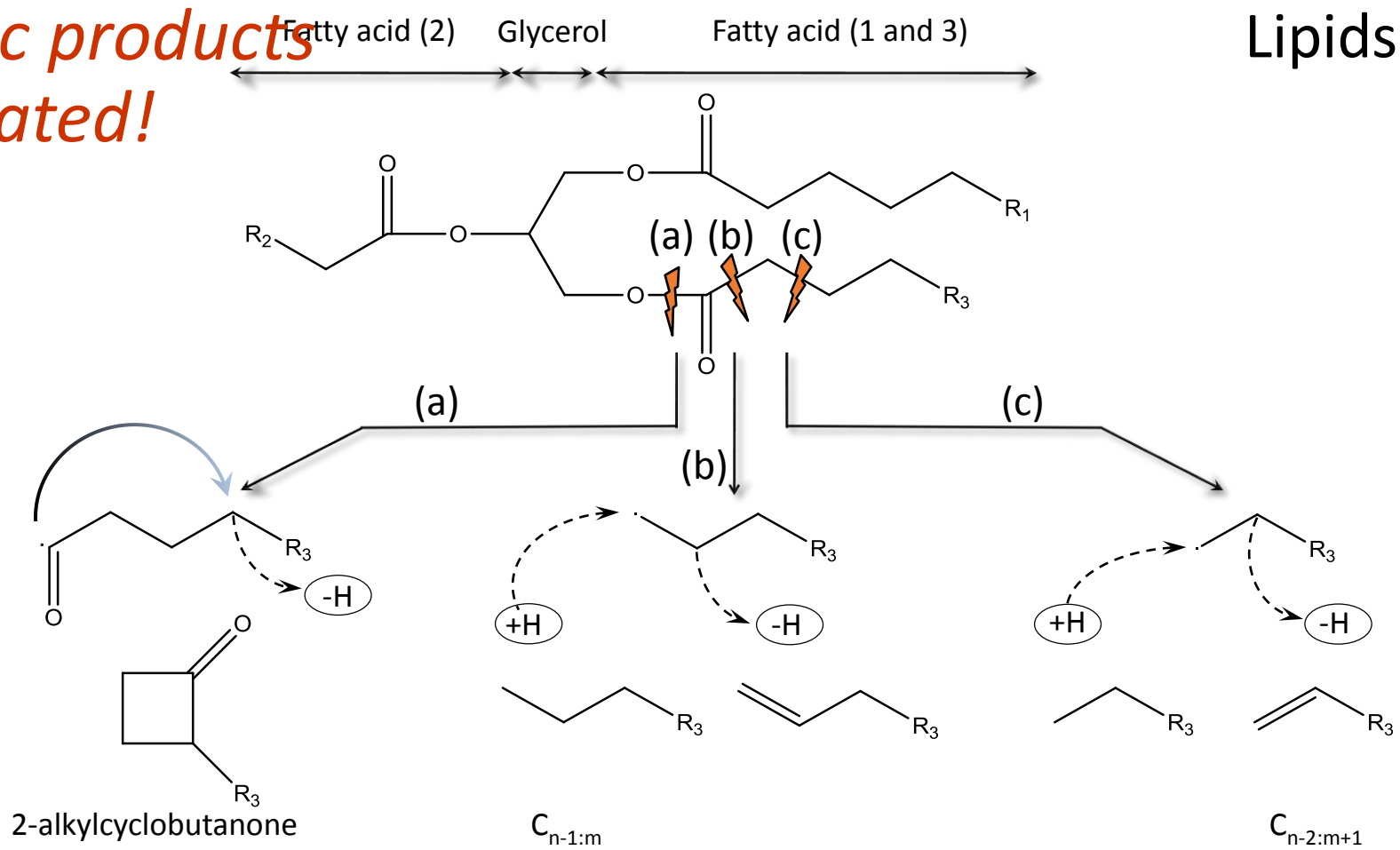
Folic acid

Radiolytic products were created!

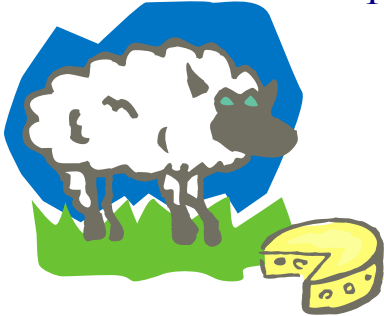
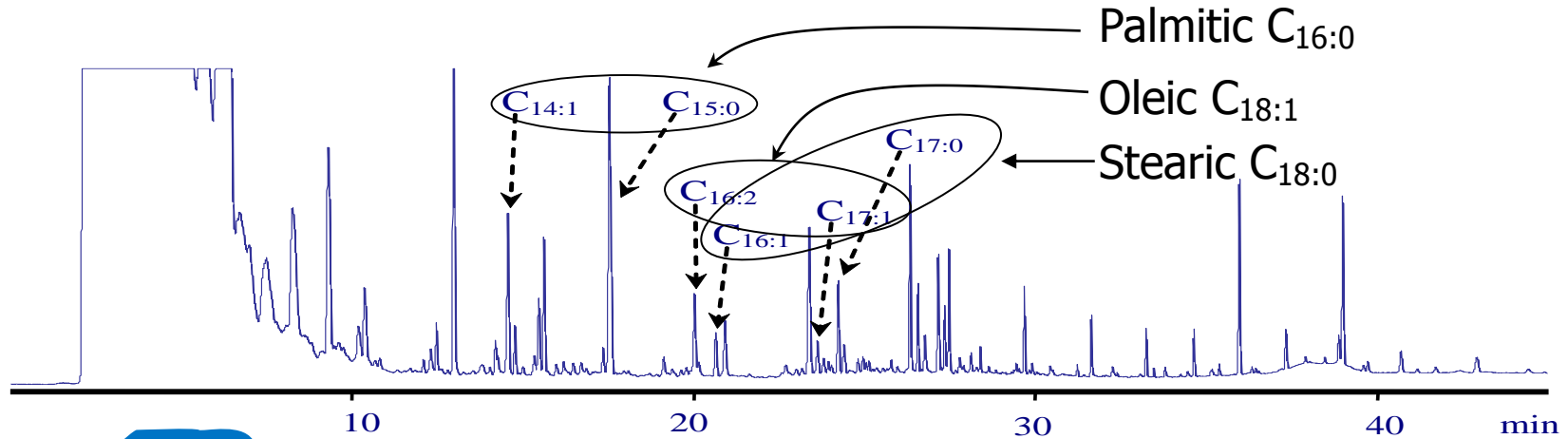
Detection UV 280 nm
Column: Agilent-XDB Phenyl
(250 x 4,6 mm; 5 μ m)
Aqueous acetic acid (0,5%)/MeOH
1,0 mL.min⁻¹
20 μ L injected



Radiolytic products were created!



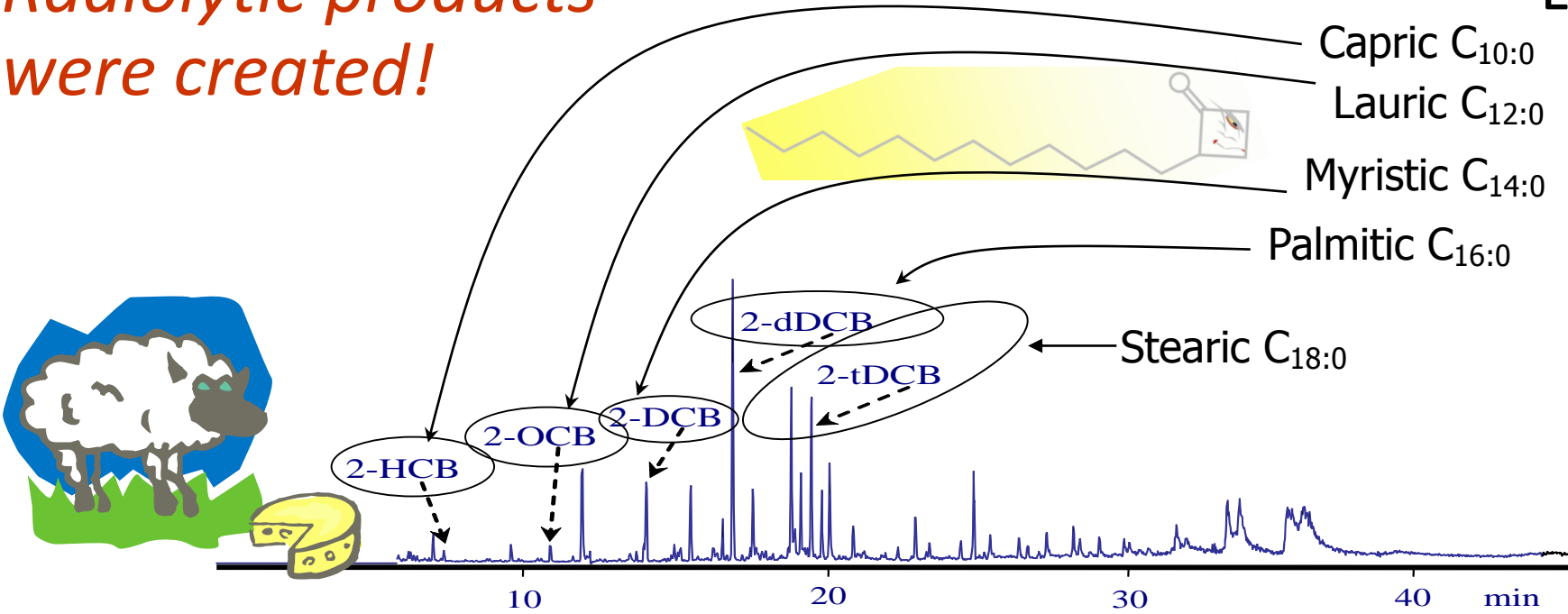
Radiolytic products were created!



Gas chromatographic analysis of volatile hydrocarbons present in a sample of cheese irradiated at 3.1 kGy (Ndiaye et al., 1999)

Radiolytic products were created!

Lipids



Gas chromatographic analysis of 2-alkylcyclobutanones in a sample of cheese irradiated at 3.1 kGy (Ndiaye et al., 1999)

*Radiolytic products
were created!*

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Natural Existence of 2-Alkylcyclobutanones

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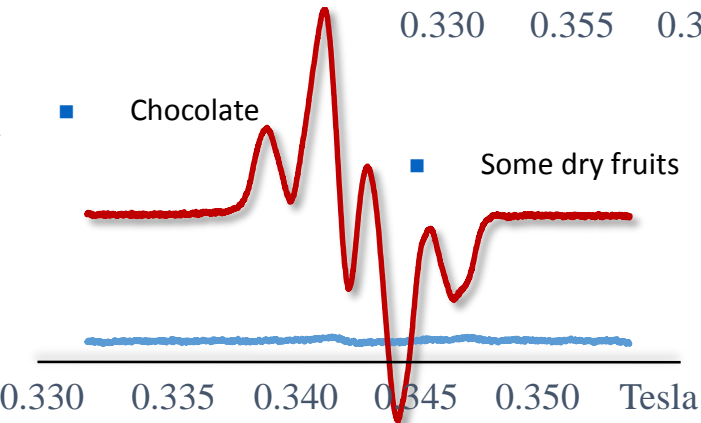
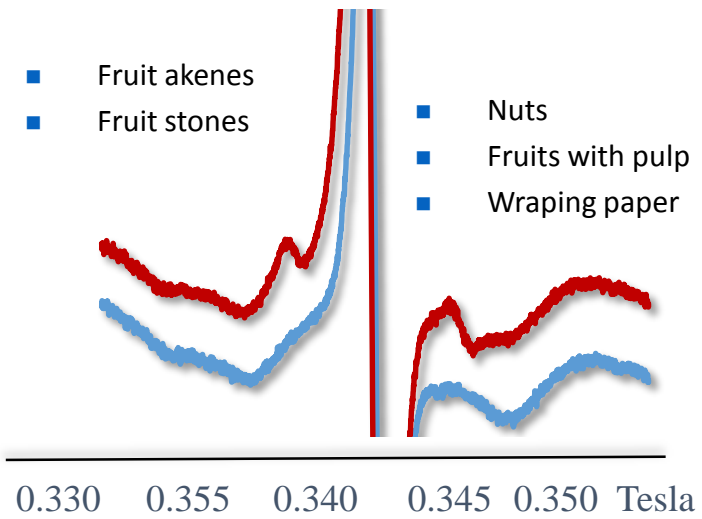
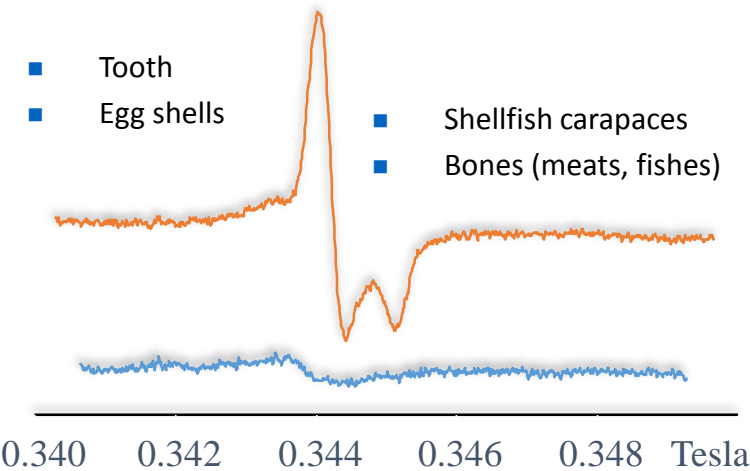
SFE-TLC-GC-MS

30 g round cashew nuts

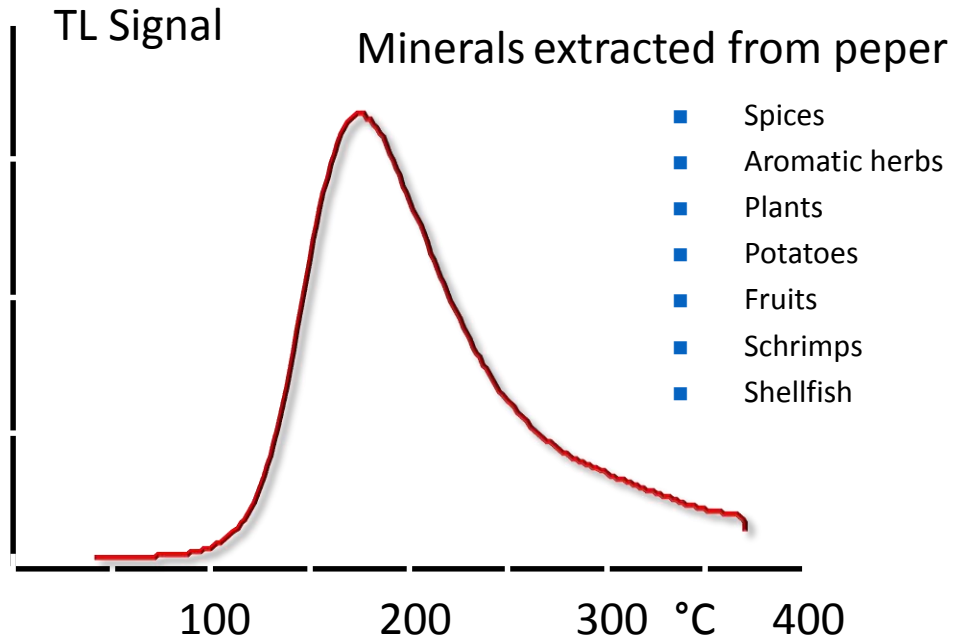
Other chemical products

- Furans
- Certain hydrocarbons
- Cholesterol oxides

Radicals were created!



Excited species were created!



Excited species were created!

**Special apparatus have
been developped to detect
these excited species in
food**

Here a PSL reader







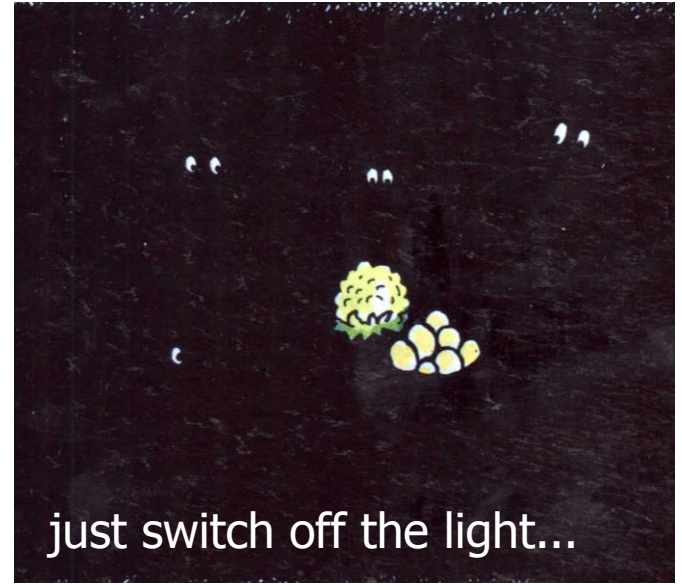
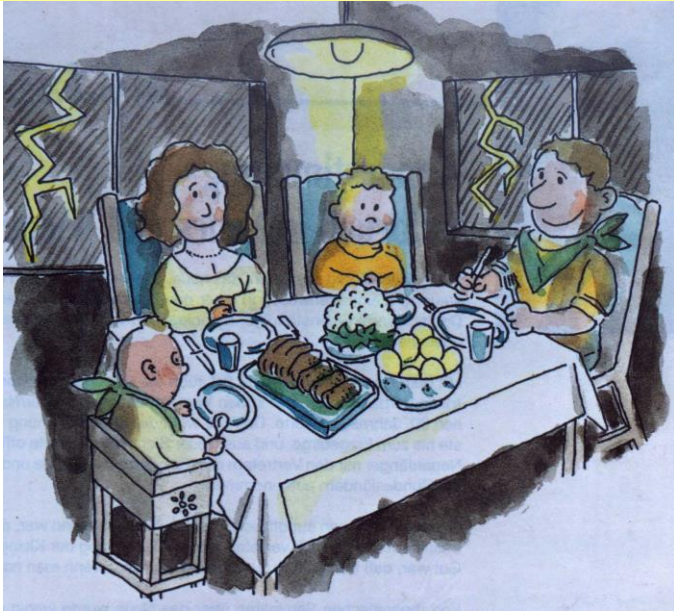
BAD TASTE



THE DISTURBING TRUTH ABOUT
THE WORLD HEALTH ORGANIZATION'S
ENDORSEMENT OF FOOD IRRADIATION

Muito obrigado pela atenção

How to detect irradiated food ?



(Der Fachberater, 1995)

Radiolytic products were created!

2-ACBs production yield : 1.6 nmole / mole fatty acid / kGy

Toxicity tests realized with very high concentrations of 2-ACBs

Toxicity tests realized with highly pure standard compounds

***In vivo* toxicity tests realized with 2-ACBs in the drinking fluid. No matrix effect was studied**

***In vivo* toxicity tests realized with rodents**

Very few metabolic studies were done (excretion, biodistribution, metabolism)